Advanced CO2 Leakage Mitigation using Engineered Biomineralized Sealing Technologies Project Number FE0004478

Lee H Spangler, Al Cunningham, Robin Gerlach Energy Research Institute Montana State University

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Presentation Outline

- Motivation
- Background information
- Large core tests ambient pressure
- Large core tests high pressure

Benefit to the Program

Program goals being addressed.

Develop technologies to demonstrate that 99 percent of injected CO_2 remains in the injection zones.

Project benefits statement.

The Engineered Biomineralized Sealing Technologies project supports Storage Program goals by developing a leakage mitigation technology for small aperture leaks that can be delivered via low viscosity solutions.

Project Overview: Goals and Objectives

The goal of this project is to develop a biomineralizationbased technology for sealing preferential flow pathways in the vicinity of injection wells.

Objective 1) Construct and test mesoscale high pressure rock test system (HPRTS).

Objective 2) Develop biomineralization seal experimental protocol.

Objective 3) Creation of biomineralization seal in different rock types and simulating different field conditions.

Target metrics for technology performance.

1) Demonstrate the ability to control the spatial distribution of the biobarrier on the 1 meter scale.

2) Achieve a 3-4 order of magnitude reduction in permeability and a 10 to 25 fold increase in capillary entry pressure.

3) Develop a barrier growth protocol consistent with field deployment

Abandoned Well Leakage Mitigation Using Biomineralization

A. Cunningham, A. Phillips, A.C. Mitchell, <u>L. Spangler</u>, and R. Gerlach Energy Research Institute Center for Biofilm Engineering **Montana State University** Bozeman MT, 59717

> Richard Esposito – Southern Company Peter Walsh – University of Alabama Birmingham



How Can We Plug Small Aperture Leaks?



Cement is a good technology for large aperture leaks, but is too viscous to plug small aperture leaks such as small fractures or interfacial delaminations

In some problematic cases it may be desirable to plug the rock formation around the well.

A missing tool is a plugging technology that can be delivered via lowviscosity fluids



How Can We Plug Small Aperture Leaks?

Approach

- Deliver materials separately in low-viscosity aqueous solutions and grow the barrier in place
- Bacteria with urealytic enzyme
- Nutrients induce biofilm growth
- **Urea Induce biomineralization**















Biomineralizing Biofilms





Engineered Applications of Ureolytic Biomineralization



Multiple Groups are Investigating Applications for Biomineralization

Figure 1. Several potential (but not limited to) engineering applications for ureolysis-driven MICP. Note: in this figure the white crystal hatch pattern represents calcium carbonate a) strengthening earthen dams or consolidating porous materials, b) application to soils to prevent dust c) remediating concrete fractures d) coating PCB-oil contaminated concrete resulting from leaky transformers e)treating or coating limestone or concrete to reduce risk of corrosive fluid infiltration f)creating ponds or reservoirs by sealing porous materials g) forming subsurface barriers to prevent unwanted fluids like salt water intrusion or contaminated groundwaters into drinking water aquifers h) remediating subsurface groundwater contaminated with radionuclides or heavy metals(represented by triangles) with co-precipitation of $CaCO_3$ i) treating fractures in cap rock to mitigate leakage from geologically sequestered carbon dioxide injection sites or coating well bore concrete to provide a sacrificial coating to prevent concrete degradation from supercritical CO₂.



Research Challenges

To move this technology forward it must be demonstrated that:

- Mineral deposits can be formed at a field relevant scale under environmental conditions appropriate to subsurface reservoirs
- Mineral deposition can be kept uniform over relevant distances (meter scale)
- The degree of sealing in disturbed rock, cement, and cement-well bore interface reaches an acceptable level
- Biomineral deposits are stable when exposed to brine/ScCO₂
- Ureolytic organisms can be isolated from the field
- Acceptable field injection protocols can be developed



High Pressure Biofilm Growth and Biomineralization Test System



- p up to 2000 psi (130 bar), T ambient to 65 C
- consolidated and unconsolidated material up to 6 in length

MITCHELL, A.C.; PHILLIPS, A.J.; HIEBERT, R.; GERLACH, R.; SPANGLER, L.; CUNNINGHAM, A.B. (2009): Biofilm enhanced geologic sequestration of supercritical CO₂. The *International Journal on Greenhouse Gas Control*. 3:90-99. doi:10.1016/j.ijggc.2008.05.002



Rock Core after Biofilm Growth & ScCO₂ Challenge





Rock Core after Biofilm Growth & ScCO₂ Challenge – SEM Images & Summary



MITCHELL ET AL. (2009) *IJGGC*. 3:90-99. doi:10.1016/j.ijggc.2008.05.002

Biofilm growth and permeability reduction at high pressure



MITCHELL, A.C.; PHILLIPS, A.J.; HIEBERT, R.; GERLACH, R.; SPANGLER, L.; CUNNINGHAM, A.B. (2009): Biofilm enhanced geologic sequestration of supercritical CO₂. The *International Journal on Greenhouse Gas Control*. 3:90-99. doi:10.1016/j.ijggc.2008.05.002



Ureolysis driven carbonate precipitation

+ pH and alkalinity (increase in OH⁻ and HCO₃⁻) increase SATURATION STATE OF CALCITE

 $CO(NH_2)_2 + H_2O \rightarrow NH_2COOH + NH_3$ $NH_2COOH + H_2O \rightarrow NH_3 + H_2CO_3$ $CO(NH_2)_2 + 2 H_2O \rightarrow 2 NH_3 + H_2CO_3 \qquad (Urea hydrolysis)$

 $2NH_3 + 2H_2O \leftrightarrow 2NH_4^+ + 2OH^-$ (pH increase)

 $H_2CO_3 + 2OH^- \leftrightarrow HCO_3^- + H_2O + OH^- \leftrightarrow CO_3^{2-} + 2H_2O$

 $CO_3^{2-} + Ca^{2+} \leftarrow \rightarrow CaCO_3$ (carbonate precipitation)

- (i) Permeability reduction
- (ii) Co-precipitation of metals
- (iii) Mineralization of CO₂



Mitchell, AC and Ferris, FG (2006) *Geomicrobiology Journal*, 23, 213-226.

Mitchell, AC and Ferris, FG (2006) Environmental Science and Technology, 40, 1008-1014.

Mitchell, AC and Ferris FG (2005) Geochimica et Cosmochimica Acta, 69, 4199-4210.



Ureolysis-Driven CaCO₃ Formation at High Pressure under Pulse-Flow Conditions





Ureolysis driven CaCO₃ formation at 89 bar under pulse-flow conditions





Ureolysis-driven CaCO₃ formation





Schultz, Pitts, Mitchell, Cunningham, Gerlach submitted to Microscopy Today



CaCO₃/Biofilm Deposits Resist Dissolution

After exposure

Before exposure



80x

Modeling of Biofilm-ScCO₂ Interactions



Ebigbo, Helmig, Cunningham, Class, Gerlach (2010) Advances in Water Resources, doi:10.1016/j.advwatres.2010.04.004



Biomineralization along a 2-foot sand column







Volume fraction of calcium carbonate-occupied pore space and $CaCO_3$ concentrations over the distance of a 60 cm column.





Biofilm Induced Calcium Carbonate Precipitation (2 ft columns)



Mesoscale Biomineralization Research

30 X 15 Inch sandstone core

Figure 1a. Sandstone core (30-inch diameter) being extracted in Alabama. *Figure 1b.* Sandstone core undergoing hydraulic testing in the MSU laboratory.



Sealing Fractures





Hydraulic testing of the 71.1 cm (28-inch) diameter Bremen core









High Pressure Vessel for 30" Cores

CAUTION

TO TADLE SLEW'A

Re-fracturing under ambient conditions after sealing under high pressure



Before Images





The core was hydraulically fractured under ambient conditions right before loading into the vessel. Distinct flow channels were formed.



STATE UNIVERSITY

During Images







After Images of fracture zone









Cell-like structures?

Mag = 5.53 K X

m

KX W

WD = 3 mm

april ---

EHT = 1.00 kV

Signal A = SE2 File Name = fracture ppt_3a.tif Date :28 Jun 201

Accomplishments to Date

- Demonstrated ability to control mineralization distribution
- Developed computational tool to simulate mineral distribution
- Successful collection of large diameter core
- Demonstrated ability to mineralize small aperture fracture under ambient pressure
- Designed and constructed high pressure vessel for large diameter core experiments
- Performed first high pressure sealing experiment on large diameter core

Summary

- Biofilm formation and biomineralization shows promise as a method to seal small aperture leaks in the subsurface
- Other mineralogy, porosity, permeability cores will be run
- Thought must be given to downhole delivery of fluids for sealing technology

Appendix

Organization Chart



Gantt Chart

Task	Description	Q1	Q2	Q3	Q4	Q5	Q6	Q7	Q8	Q9	Q10	Q11	Q12
	- · · · ·	←											\rightarrow
1	Project Management & Planning												-
2	Construction of high pressure rock testing systems (HPRTS)	←					\rightarrow						
~	construction of high pressure rock testing systems (in Kroj												
2.1	Design and fabricate HPRTS system	<					\rightarrow						
				4			_						
2.2	Initial testing of HPRTS						-						
		←					\rightarrow						
2.3	Charactering the initial flow properties of rock samples	-					-						
2	Develop higmineralization seal experimental protocol						\rightarrow						
	Develop bioinneralization sear experimental protocol												
3.1	Radial Flow				<		\rightarrow						
					4								
3.2	Axial (Linear) Flow						-						
					-		\rightarrow						
3.3	Assessment of effectiveness of biomineralization seal						-						
л	creation of biomineralization seal in different rock types					-							\rightarrow
4													
4.1	Additional Experiments					-							\rightarrow
											_		
4.2	ScCO2 challenges of mineralized rock										-		
			4										
5	Experimental Simulation Modeling of Processes												-
E 1	Pro experimental modeling		+										\rightarrow
5.1	rre-experimental modeling												
5.2	Post-experimental modeling		+										\rightarrow

Bibliography

• There have been three presentations, one patent, and one computer simulation model produced from this project. A paper is being prepared for submission.